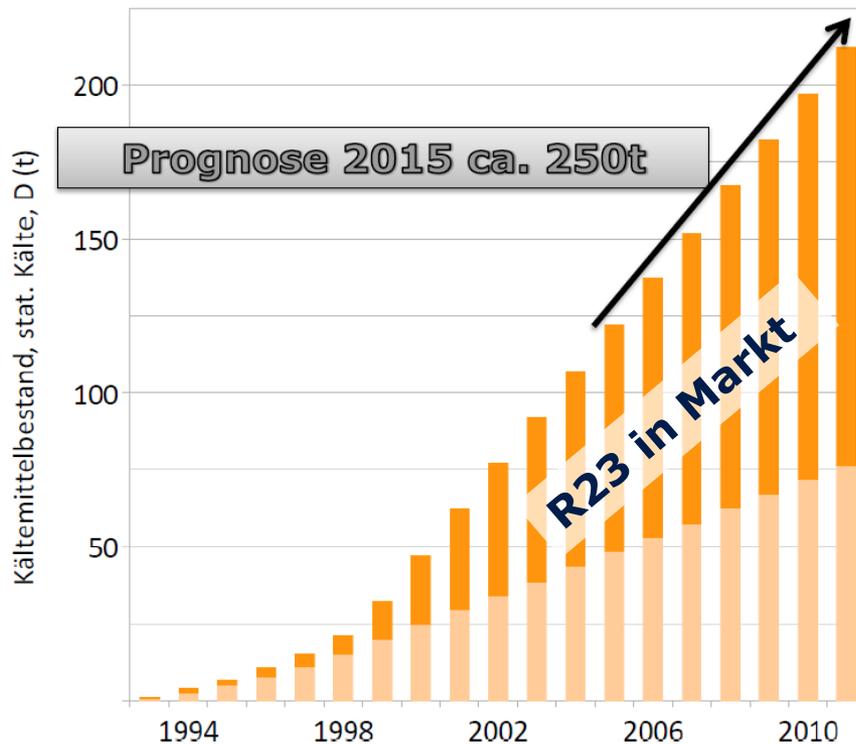




Assessment of alternatives to R23

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R23 Mengenanalyse – Studie „Kältetechnologien Deutschland“ (HsKA)



Demand in Germany

250 t (GWP R23 = 14.800)
 → 3,7 Mt CO₂ per year

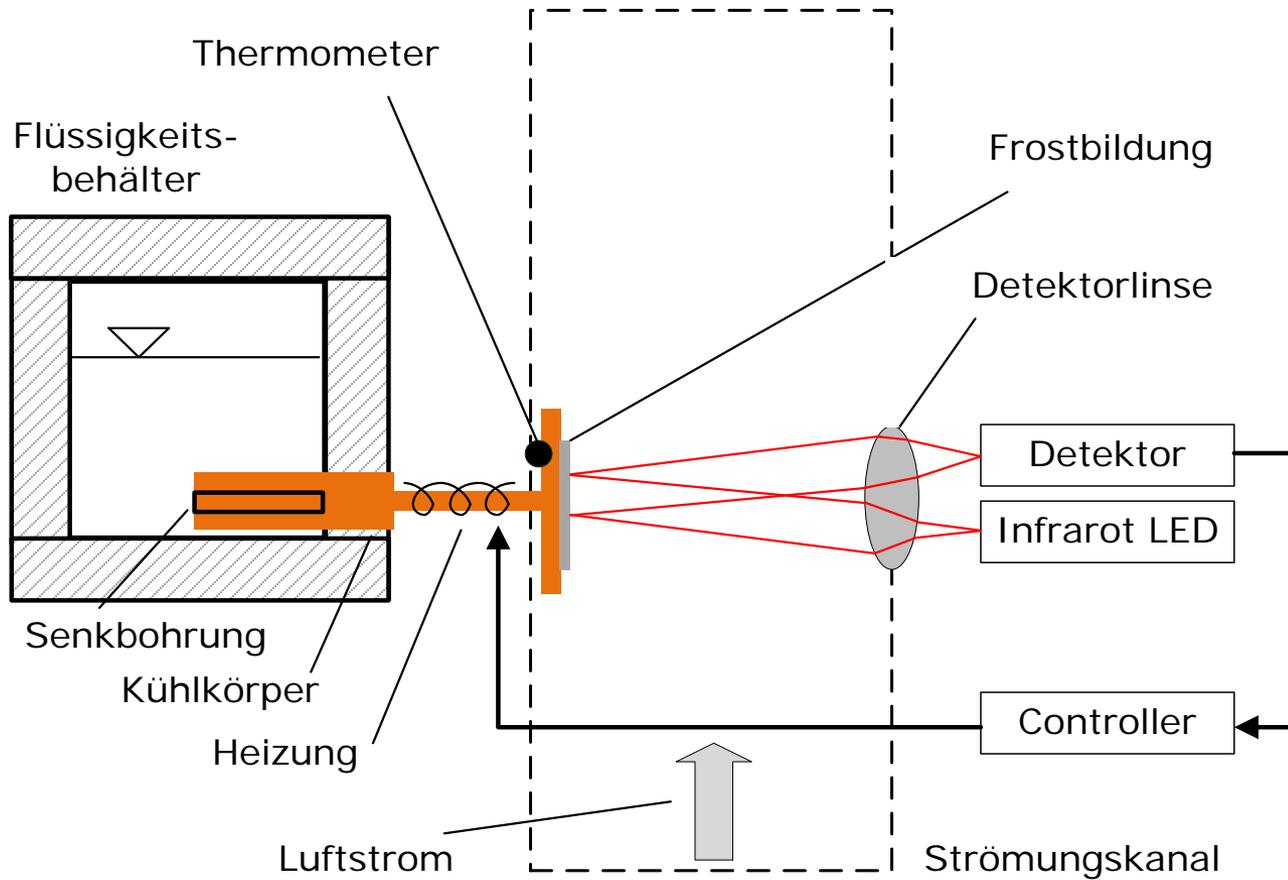
Price development

from 7 €/t_{CO2} to > 35 €/t_{CO2}

March 2018:

R23: 100 ... 300 € / kg

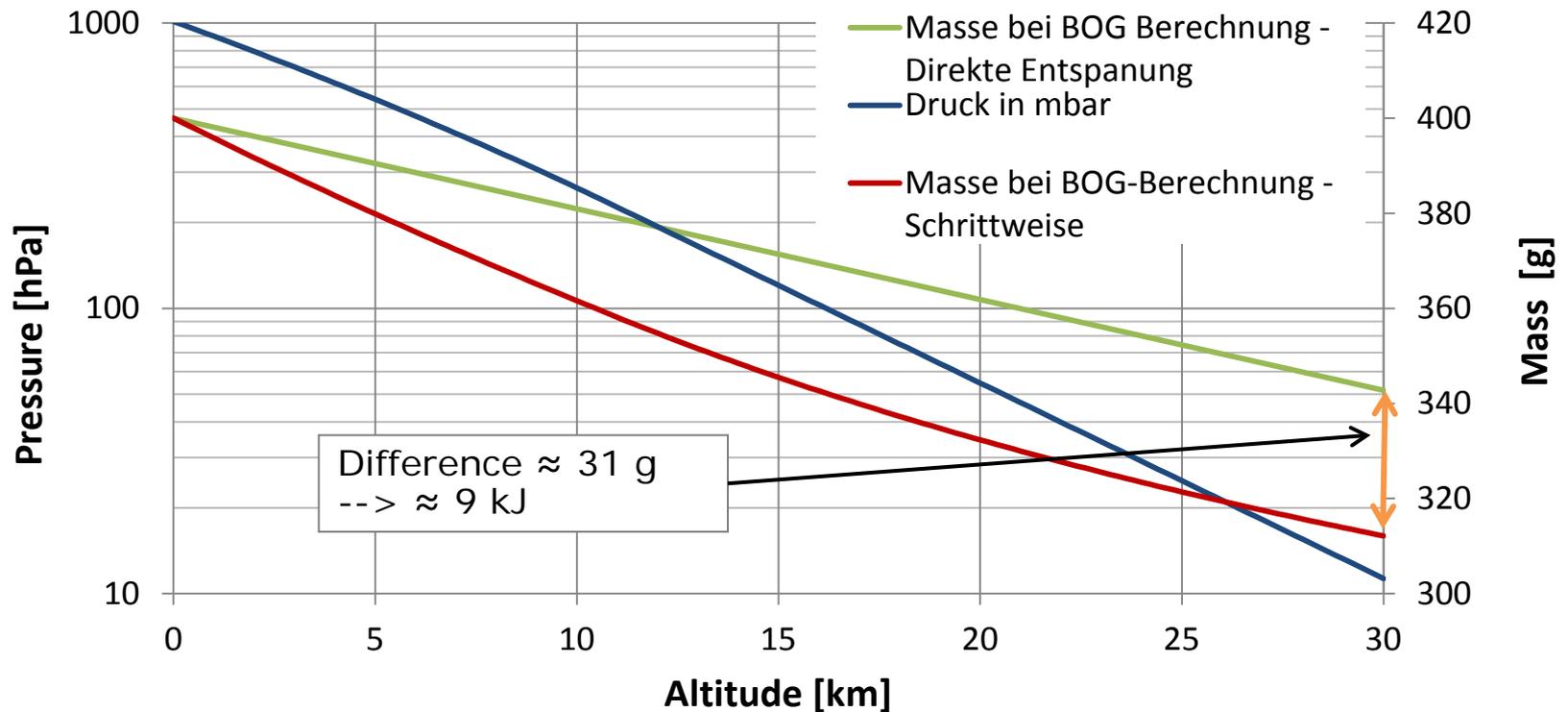
FH operating principle



Estimate for a typical sounding

400 g R23 for 3 hr of CFH operation $\rightarrow \approx 117 \text{ kJ} \rightarrow 10,5 \text{ W}$

Losses due to „Boil-off“ (1000 hPa at surface, 11 hPa at 30 km)



Technical constraints

Quantity	Value
Amount of R23 for 3h sounding [kg]	0,4
Max payload mass [kg]	4,5
Actual payload mass [kg]	2,5
Average cooling power [W]	10,5
Max cooling power [W]	12
Max mass density in [g/cm ²]	13

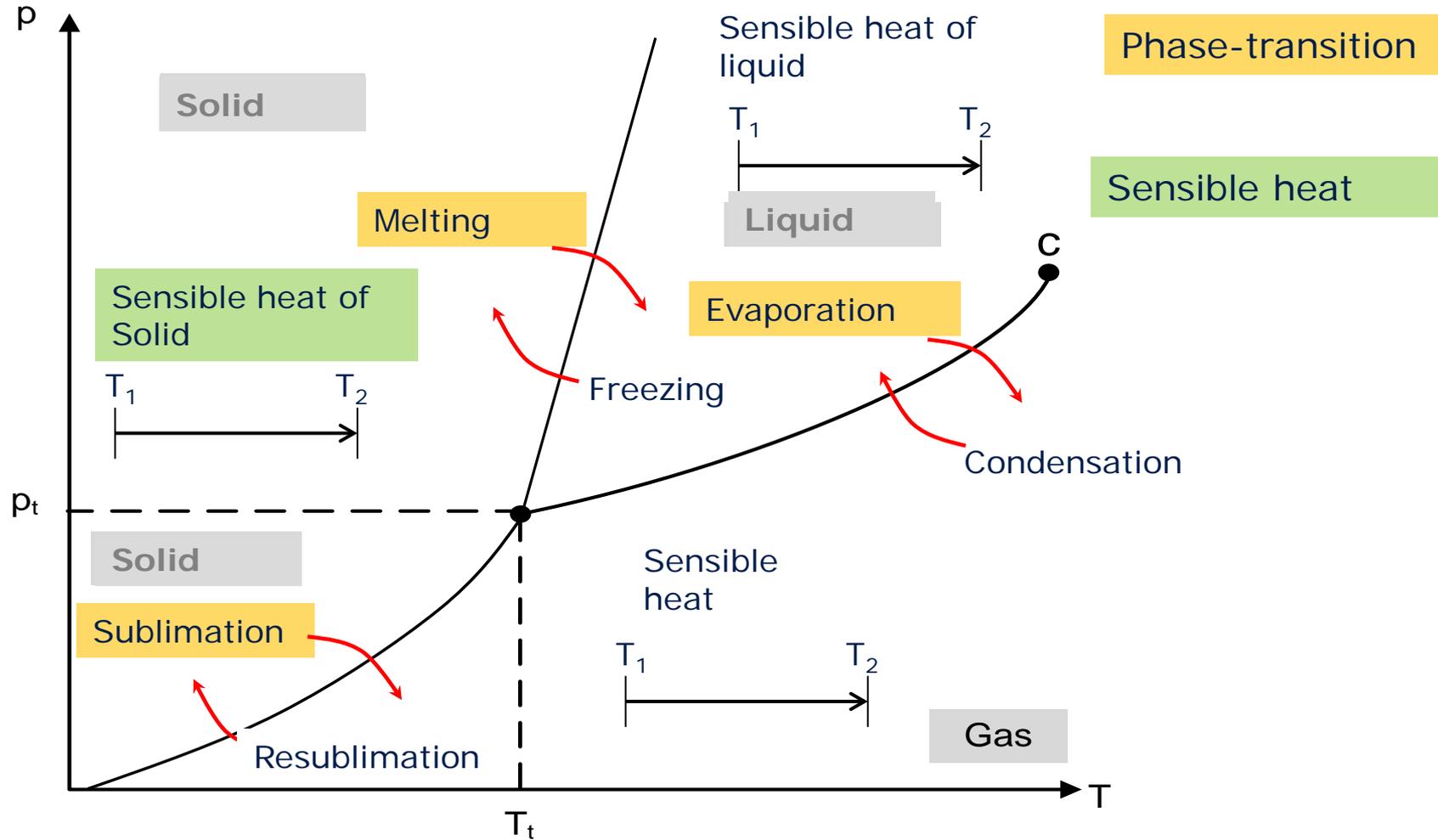
Physical constraints

Quantity	Value
Temperature at 1000 hPa in [K]	< 193
Temperature at 100 hPa [K]	< 178
Temperature at 5 hPa [K]	≈ 132
Triplepoint pressure [hPa]	< 5
Triplepoint temperature [K]	< 140

Cryogen has to be available commercially

Non toxic, -flammable, -corrosive/oxydative

Question: Which cooling mechanisms can be applied?



Mechanisms:

Baro-, magneto- and elasto- and Electrocalorimetry

Challenge (constraints):

Temperature difference: > 50 K

Power: up to 12 W

Mass limit: 2 kg

→ No technical solution available

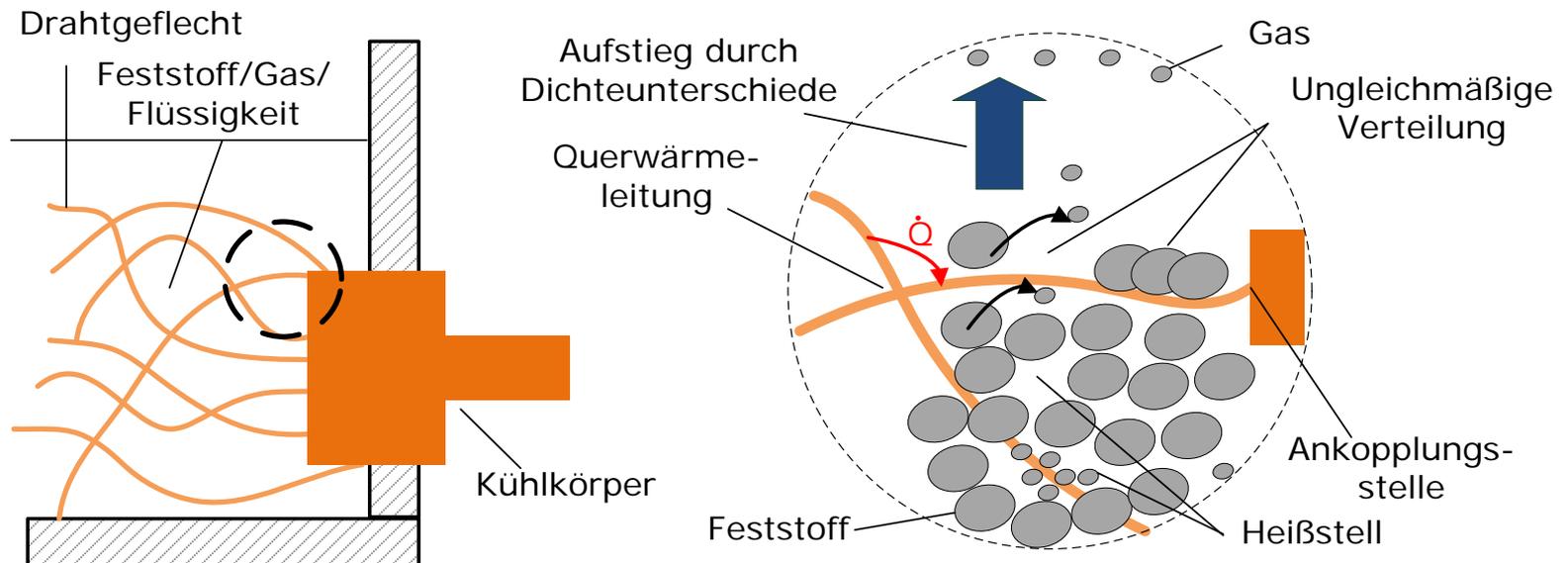
Evaporation:

Heat: 113 kJ \rightarrow 0,5 kg (current design) \rightarrow 230 kJ / kg

Melting:

Heat: 113 kJ \rightarrow 2,4 kg (Maximum) \rightarrow 50 kJ / kg

Sublimation: Heat capacity?



Evaporation: (Nr of assessed fluids: ≈2000)

Fluid	NBP [K]	T_{trip} [K]	p_{trip} [hPa]	T_{boil} @ 5 hPa [K]	Issue
Ethane	184,6	90,4	0,01	122,5	Flammable
Ethene	169,4	104,0	1,22	112,7	Flammable
Oxygen	90,2	54,4	1,46	58,6	Oxidative
Fluor	85,0	53,5	2,39	55,9	Reaktive, toxic
COS	223,0	134,3	0,64	150,1	Flammable, toxic
R32					Flammable,
	221,5	136,3	0,48	153,5	GWP = 650
R41	194,8	129,8	3,45	132,6	Flammable
R1141	201,0	113,0	0,34	132,5	Flammable

NBP = Normal boiling point (@ 1000 hPa)

Conclusion: No alternative that satisfies all criteria!

Melting :

Fluid: ($T_{\text{melt}} < 153\text{K}$)

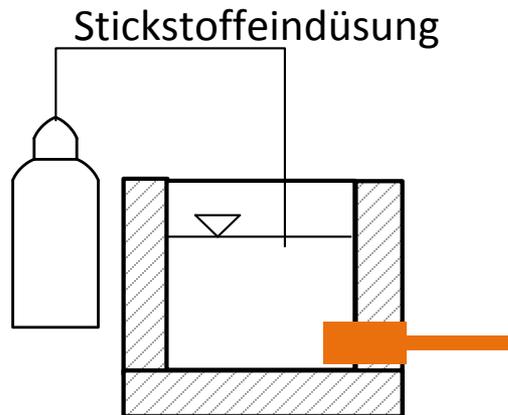
Butane, Pentane, Hexan (Iso- and Normalforms)
Ethers

Fluid: ($T_{\text{melt}} < 173\text{K}$)

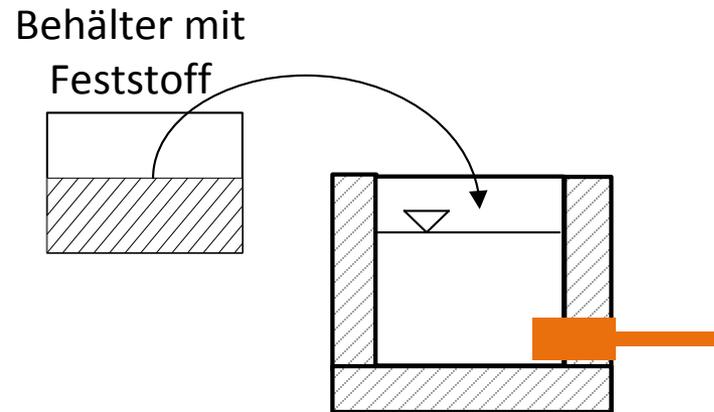
Alcohol, Polyflourether
Hydrofluorocarbon

Fluid	R134a	R245fa	R1234ze
kJ / kg	25	60	95

Feststoffherzeugung in Apparatur



Feststoffherzeugung Extern



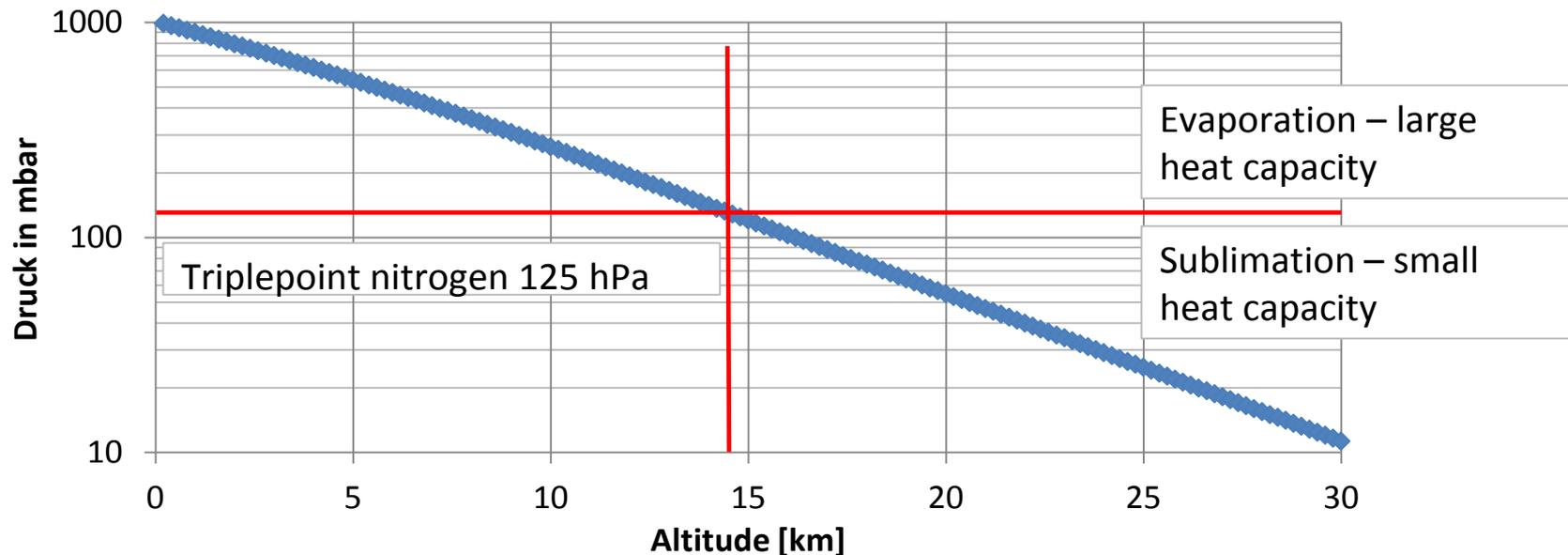
Sublimation:

Fluids: ($T_{\text{sub}} < 153\text{K}$) Cryogenic (Ar, N₂, CH₄)

Fluids: ($T_{\text{sub}} < 173\text{K}$) Hydrofluorocarbons (R125, R1123 ...)

Fluid	Ar	N ₂	CH ₄
kJ / kg	195	262	604

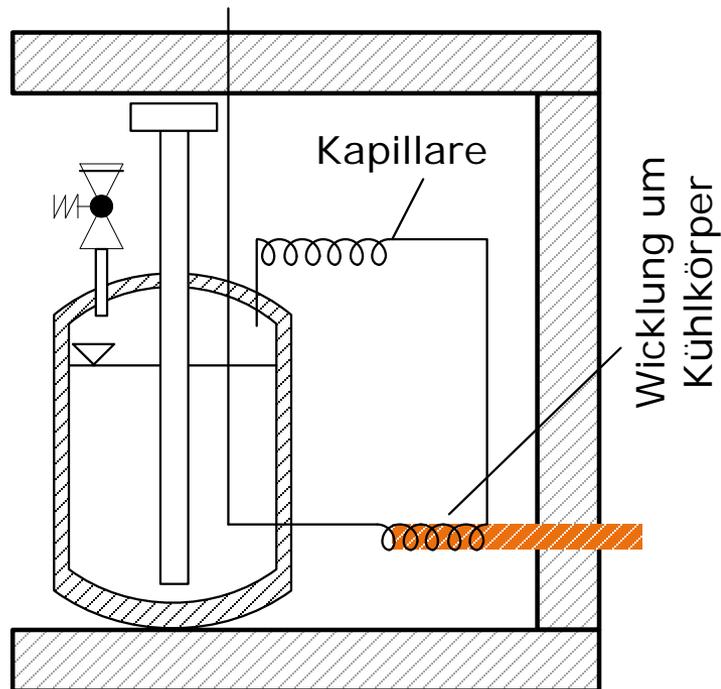
Example: N₂



Problem: Triplepoint N₂ 125 hPa >> p_{min} = 5(10) hPa Sublimation

Idea: Liquid N₂ in pressurized Dewar (0.83 l N₂ ≅ 0.66 kg)

Cooling with cold nitrogen gas



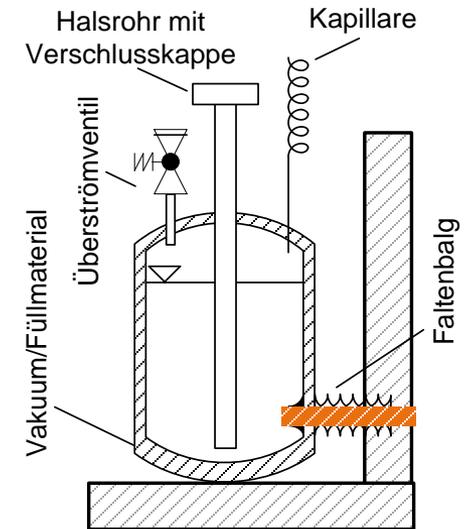
Nutzung Kaltgas mit
Gasrückführung zur Druckhaltung

Approach:

Pressure retention in Dewar by capillary

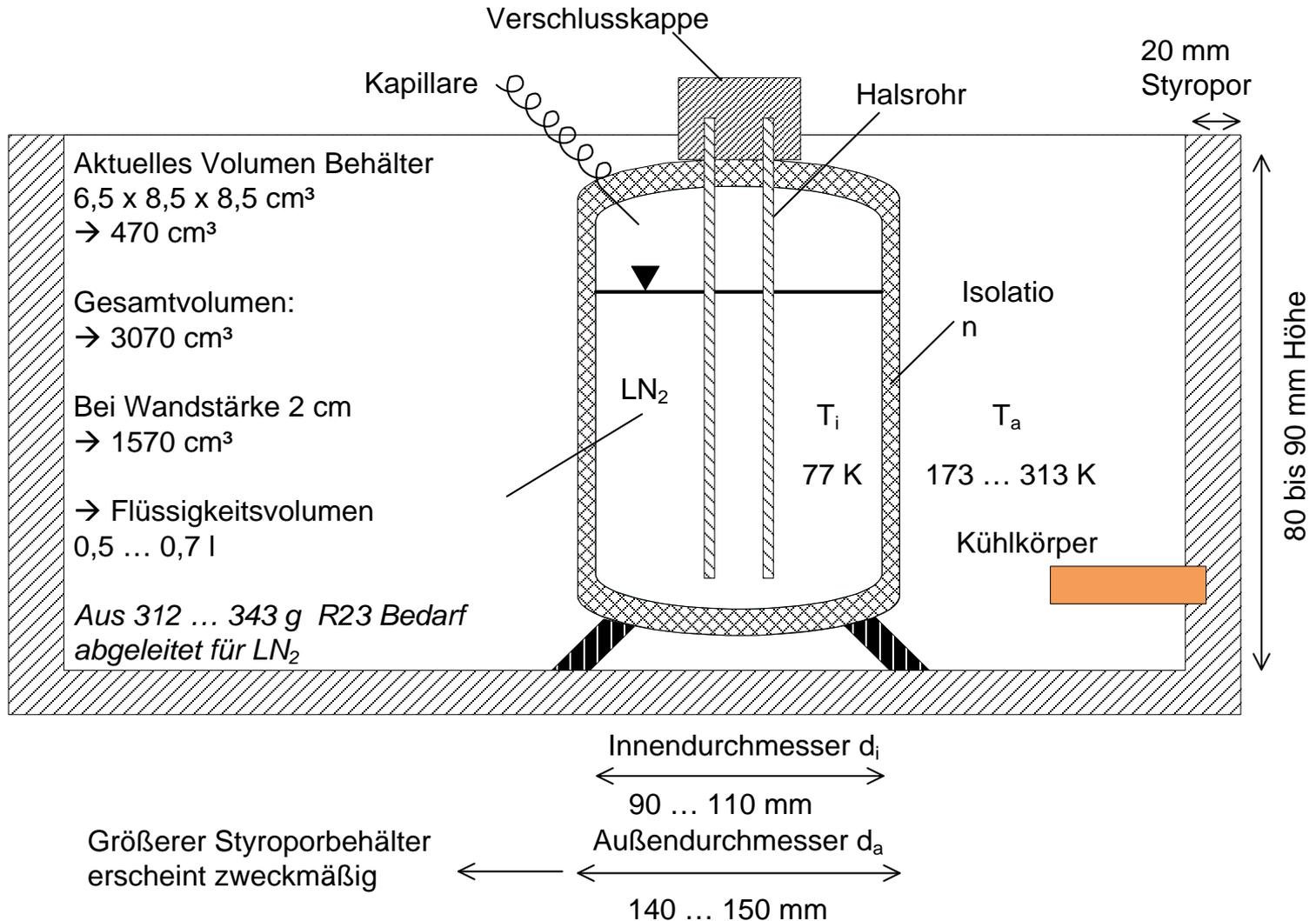
Parameters for capillaries:

Length 1 m
Internal diameter: 0,7 mm



Direkte Anbindung

Pressure[bar]	Exit pressure [hPa]	Mass flow [kg/h]	Cooling power [W]	Remarks
2,2	1000	0,27	15	-
2,0	5	0,27	15	Choke limit of capillary
1,4	5	0,18	10	Choke limit of capillary
0,7	5	0,09	5	Choke limit of capillary
0,2	5	0,025	1,5	minimal pressure to sustain flow



- Need for alternative to R23 as cryogen for frostpoint hygrometers
- Preferred cooling mechanisms: Evaporation, melting, sublimation
- A fluid is available for all transitions/mechanisms
- Sublimation: small heat capacity
- No alternative to R23 that matches required properties
- Alternative option: Liquid N₂ in Dewar with capillary for flow control (choke limit)

Next step: Testing of

- sublimation + Metal mesh
- Dewar + Capillary

Critical questions ?

